

Total Synthesis of 1,3-Dideoxynojirimycin

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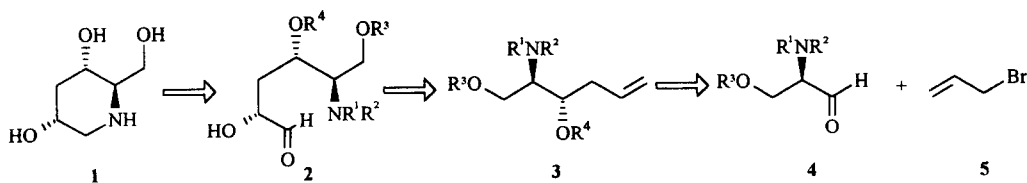
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Abstract: Addition of allyl bromide (5) to *N*-benzyl-*N*-carbobenzoxy-*O*-*tert*-butyldimethyl-D-serinal (4) afforded with high diastereoselectivity *anti*-adduct 6 which subsequently transformed into (2*R*,4*S*,5*R*)-1,3-dideoxynojirimycin (1). © 1997 Elsevier Science Ltd.

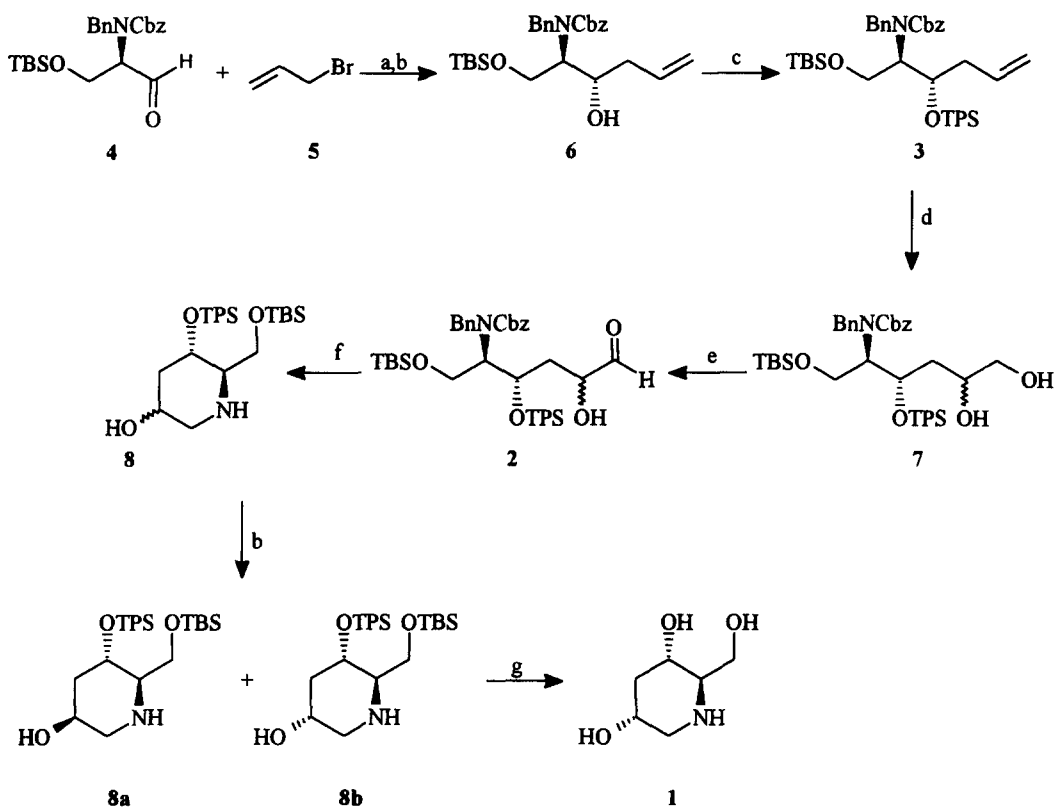
Polyhydroxypiperidines constitute an interesting group of biologically active compounds of potential pharmacological importance, since they are effective inhibitors of glycosidases.^{1,2} The known methods of their syntheses are mainly based on transformation of naturally occurring D-pentoses or D-hexoses.³⁻⁵ Polyhydroxypiperidines can also be synthesized from nonsugar precursors. For example, Johnson *et al.*⁶ have recently published a new method of the synthesis of 1,3-dideoxynojirimycin (1) using an enzymatic dissymetrization of cyclopentadiene.

During our studies on applications of *N*-protected α -amino aldehydes in organic synthesis, we have found that they are very convenient, versatile and effective chiroins.⁷⁻⁹ Now we report a new application of our methodology to the synthesis of (2*R*,4*S*,5*R*)-1,3-dideoxynojirimycin (1). Retrosynthetic analysis, shown in Scheme 1, suggested that *N*-benzyl-*N*-carbobenzoxy-*O*-*tert*-butyldimethylsilyl-D-serinal (4)^{10,11} and allyl bromide (5) could serve as starting materials.

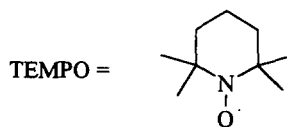
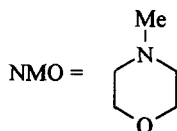
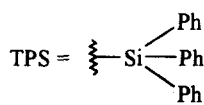
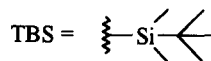
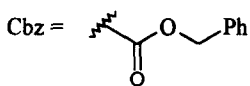


Scheme 1

Addition of allyl bromide (5) to aldehyde 4, in the presence of $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$ and sodium iodide in DMF at room temperature,^{17,18} afforded with high diastereoselectivity (89:11) *anti*-adduct 6¹⁹ in 81% yield (Scheme 2).



Scheme 2. Reagents and reaction conditions: (a) $\text{SnCl}_2 \cdot 2\text{H}_2\text{O}$, NaI, DMF, RT; (b) chromatographic separation; (c) TPSCl, pyridine, CH_2Cl_2 , RT; (d) NMO, OsO_4 , *tert*-BuOH, H_2O , THF, RT; (e) NaOCl, TEMPO, NaBr, toluene, AcOEt, H_2O , 0°C ; (f) H_2 , Pd/C, MeOH, RT; (g) Ref. 26.



Chromatographically pure *anti*-**6** was *O*-protected with TPS group²² and the resulting compound *anti*-**3** (80% yield) was *syn*-dihydroxylated²³ to afford in 90% yield diol **7** as a 1:1 diastereoisomeric mixture. This mixture was regiospecifically oxidized using the TEMPO method,^{15,16} and the resulting aldehyde **2** was subjected to hydrogenation in the presence of catalytic amounts of palladium-on-charcoal (Degussa), affording a 1:1 diastereoisomeric mixture of cyclization product **8** (75% overall yield, starting from **7**). The chromatographic separation of this mixture yielded both diastereoisomers **8a** and **8b** (less polar) in optically pure form.²⁴ The configuration (*2R,4S,5R*) was assigned to the less polar diastereoisomer **8b** on the basis of correlation ¹H-¹H and ¹H-¹³C NMR spectra.²⁵ The final transformation of compound **8b** into 1,3-dideoxynojirimycin (**1**) could be easily achieved by simple removal of both silyl protecting groups.^{26,27}

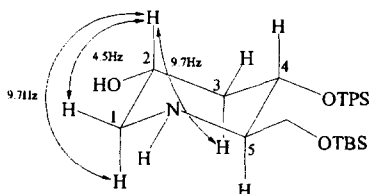
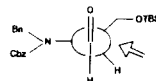
The presented total synthesis of 1,3-dideoxynojirimycin (**1**) proves to be a practical alternative to the known procedure.³⁻⁶

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REFERENCES AND NOTES

1. Truscheit, W.; Frommer, B.; Junge, L.; Muller, L.; Schmidt, D.D.; Wingender, W. *Angew. Chem. Int. Ed. Engl.* **1981**, *20*, 755.
2. Fleet, G. W. *Chem. Brit.* **1989**, 287.
3. Fleet, G. W.; Carpenter, N. M.; Petursson, S.; Ramsden, S. G. *Tetrahedron Lett.* **1990**, *31* 409.
4. Baxter, E. W.; Reitz, A. *Bioorg. Med. Chem. Lett.* **1992**, *2*, 1419.
5. Hudlicky, T.; Rouden, J.; Luna, H. *J. Org. Chem.* **1993**, *58*, 985.
6. Johnson, C. R.; Gołębiowski, A.; Braun, M. P.; Sundram, H. *Tetrahedron Lett.* **1994**, *35*, 1833.
7. Jurczak, J.; Gołębiowski, A. *Chem. Rev.* **1989**, *89*, 149.
8. Gołębiowski, A.; Jurczak, J. *Synlett* **1993**, 241.
9. Kiciak, K.; Jacobsson, U.; Gołębiowski, A.; Jurczak, J. *Polish J. Chem.* **1994**, *68*, 199.
10. Jurczak, J.; Gryko, D.; Kobrzycka, E.; Gruza, H.; Prokopowicz, P. *Tetrahedron*, submitted.
11. The D-serinal derivative **4** was obtained as follows: *N*-benzyl-D-serine methyl ester was treated with benzyl chloroformate in the two-phase system,¹² affording *N*-Bn-*N*-Cbz-D-serine methyl ester (92% yield), which was then reacted with *tert*-butyldimethylsilyl chloride in the presence of imidazole in CH₂Cl₂/DMF,¹³ giving *N*-Bn-*N*-Cbz-*O*-TBS-D-serine methyl ester (98% yield), which was reduced with LiBH₄ in THF/EtOH,¹⁴ affording the corresponding alcohol (81% yield). Oxidation of the alcohol using the TEMPO procedure^{15,16} gave aldehyde **4** (90% yield), which did not require further purification.

12. Gołębiowski, A.; Gorins, G.; Johnson, C. R.; Kiciak, K. *Polish J. Chem.* **1993**, *67*, 685.
13. Gołębiowski, A.; Raczko, J.; Jurczak, J. *Bull. Pol. Ac. Chem.* **1988**, *36*, 209.
14. Hamada, Y.; Shioiri, T. *J. Tetrahedron Lett.* **1982**, *23*, 1193.
15. Leanna, M. R.; Sowin, T. J.; Morton, H. E. *Tetrahedron Lett.* **1992**, *33*, 5029.
16. Jurczak, J.; Prokopowicz, P.; Gołębiowski, A. *Tetrahedron Lett.* **1993**, *34*, 7107.
17. Imai, T.; Nishida, S. *Synthesis* **1993**, 395.
18. Rüdsum, F.; Seck, S.; Giannis, A. *Tetrahedron* **1997**, *53*, 2823.
19. The stereochemical results of addition can be rationalized by the transition state model as proposed by Felkin *et al.*²⁰ and modified by Anh.²¹
20. Cherest, M.; Felkin, H.; Prudent, N. *Tetrahedron Lett.* **1968**, *18*, 2199.
21. Anh, N. T. *Top. Curr. Chem.* **1980**, *88*, 114.
22. Nakai, H.; Hayashi, M. *Chem. Lett.*, **1979**, 1499.
23. Jarosz, S. *Polish J. Chem.* **1992**, *66*, 1853.
24. Satisfactory analyses and spectral data were obtained for all new compounds.
25. The signal originated from the proton H-2 is a doublet of doublets of triplets and exhibits three coupling constants $J_1=J_2=9.7$ and $J_3=4.5$ Hz, typical for axial-axial-equatorial interactions:



26. Corey, E. J.; Venkateswarlu, A. *J. Am. Chem. Soc.* **1972**, *94*, 6190.
27. Analytical data for **1**: mp 191-193°C (decomp.), [Lit.⁶ mp 192-193°C (decomp.)], $[\alpha]_D^{20} +22.5$ (*c* 0.3, H₂O), [Lit.⁶ $[\alpha]_D^{20} +24.9$ (*c* 0.5, H₂O)]; ¹H and ¹³C NMR spectra were identical with those published in Ref.6.

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